

# Toward Tailoring the Specificity of the S<sub>1</sub> Pocket of Subtilisin *B. lentus*: Chemical Modification of Mutant Enzymes as a Strategy for Removing Specificity Limitations<sup>†</sup>

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**ABSTRACT:** In both protein chemistry studies and organic synthesis applications, it is desirable to have available a toolbox of inexpensive proteases with high selectivity and diverse substrate preferences. Toward this goal, we have generated a series of chemically modified mutant enzymes (CMMs) of subtilisin *B. lentus* (SBL) possessing expanded S<sub>1</sub> pocket specificity. Wild-type SBL shows a marked preference for substrates with large hydrophobic P<sub>1</sub> residues, such as the large Phe P<sub>1</sub> residue of the standard suc-AAPF-pNA substrate. To confer more universal P<sub>1</sub> specificity on S<sub>1</sub>, a strategy of chemical modification in combination with site-directed mutagenesis was applied. For example, WT-SBL does not readily accept small uncharged P<sub>1</sub> residues such as the –CH<sub>3</sub> side chain of alanine. Accordingly, with a view to creating a S<sub>1</sub> pocket that would be of reduced volume providing a better fit for small P<sub>1</sub> side chains, a large cyclohexyl group was introduced by the CMM approach at position S166C with the aim of partially filling up the S<sub>1</sub> pocket. The S166C-S-CH<sub>2</sub>-c-C<sub>6</sub>H<sub>11</sub> CMM thus created showed a 2-fold improvement in  $k_{\text{cat}}/K_{\text{M}}$  with the suc-AAPA-pNA substrate and a 51-fold improvement in suc-AAPA-pNA/suc-AAPF-pNA selectivity relative to WT-SBL. Furthermore, WT-SBL does not readily accept positively or negatively charged P<sub>1</sub> residues. Therefore, to improve SBL's specificity toward positively and negatively charged P<sub>1</sub> residues, we applied the CMM methodology to introduce complementary negatively and positively charged groups, respectively, at position S166C in S<sub>1</sub>. A series of mono-, di-, and trinegatively charged CMMs were generated and all showed improved  $k_{\text{cat}}/K_{\text{M}}$ s with the positively charged P<sub>1</sub> residue containing substrate, suc-AAPR-pNA. Furthermore, virtually arithmetic improvements in  $k_{\text{cat}}/K_{\text{M}}$  were exhibited with increasing number of negative charges on the S166C-R side chain. These increases culminated in a 9-fold improvement in  $k_{\text{cat}}/K_{\text{M}}$  for the suc-AAPR-pNA substrate and a 61-fold improvement in suc-AAPR-pNA/suc-AAPF-pNA selectivity compared to WT-SBL for the trinegatively charged S166C-S-CH<sub>2</sub>CH<sub>2</sub>C(COO<sup>–</sup>)<sub>3</sub> CMM. Conversely, the positively charged S166C-S-CH<sub>2</sub>CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> CMM generated showed a 19-fold improvement in  $k_{\text{cat}}/K_{\text{M}}$  for the suc-AAPE-pNA substrate and a 54-fold improvement in suc-AAPE-pNA/suc-AAPF-pNA selectivity relative to WT-SBL.

For both protein chemistry (1–3) and organic synthesis applications (4–6), it is desirable to have available a diverse toolbox of inexpensive proteases with high selectivity and diverse substrate preferences. To date, the most extensively exploited class of enzymes in organic synthesis applications has been the hydrolases. Among these, the serine proteases have received considerable attention due, in part, to their often exquisite stereo-, regio-, and chemoselectivities (4–

7). While over 3000 enzymes have now been reported, of which many are proteases, significantly fewer of the latter are available inexpensively from commercial sources (5, 8). Furthermore, since WT<sup>1</sup> enzymes will never accept all substrate structures of synthetic interest, it is attractive to contemplate the tailoring of a readily available protease in order to expand its substrate specificity in a controlled manner with the ultimate goal of creating any desired specificity at will.

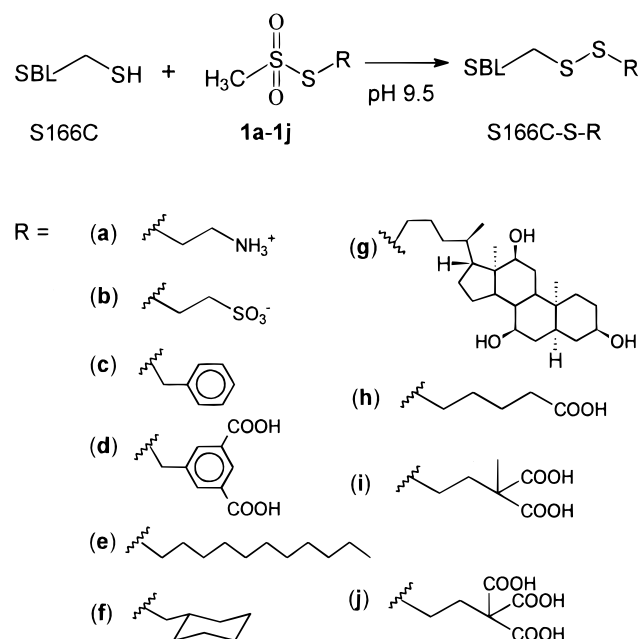
In this regard, the goal of specificity alteration of enzymes has already been targeted by several different approaches. For example, site-directed mutagenesis (9) and random mutagenesis (10) have been employed to tailor enzyme specificity and have permitted some insights into the electrostatic (11–16), steric (17–22), and hydrophobic (20, 23–24) factors which govern enzyme–substrate interactions. In particular, the directed evolution approach to altering enzyme specificity is now beginning to make significant advances toward these goals (10). However, the structural variations within these approaches are limited to the 20

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<sup>1</sup> Abbreviations: Ches, 2-(cyclohexylamino)ethanesulfonic acid; CMM, chemically modified mutant enzyme; DMSO, dimethyl sulfoxide; ES-MS, electrospray mass spectrometry; EtOH, ethanol; FPLC, fast-performance liquid chromatography; Mes, 4-morpholineethanesulfonic acid; MTS, methanethiosulfonate; PAGE, polyacrylamide gel electrophoresis; PEG, poly(ethylene glycol); SBL, subtilisin *Bacillus lentus*; SDM, site-directed mutagenesis; Suc-AAPF-pNA, succinyl-alanyl-alanyl-prolyl-phenylalanyl-para-nitroanilide; Tris, tri(hydroxymethyl)aminomethane; TFA, trifluoroacetic acid; WT, wild-type.

Scheme 1



natural amino acids. Consequently, biosynthetic methods have recently been developed to introduce unnatural amino acids into proteins (25–27). Unnatural functionalities have also been incorporated by chemical modification techniques (28–34). Since the unnatural amino acid mutagenesis approach is not yet amenable to large-scale preparations and chemical modification alone is insufficiently specific, we have begun to exploit a strategy of applying a combination of site-directed mutagenesis and chemical modification to modify enzyme specificity (35–40). This approach is illustrated in Scheme 1 and entails the introduction of a unique cysteine residue at a selected position, followed by its chemical modification with methanethiosulfonate (41–43) reagents (MTS, **1a–j**) to generate chemically modified mutant enzymes (CMMs). The combination of site-directed mutagenesis and chemical modification has previously been recognized as a powerful tool for the creation of new active-site environments (19, 44), in mechanistic studies (45–46) for the investigation of protein packing (47), and for cofactor incorporation (28). This approach has also been applied to detailed studies of ion-channel properties (48–50) for site-directed introduction of spin labels (51, 52), to probe receptor binding (53), and in investigations of membrane spanning proteins (54, 55).

The subtilisin from *Bacillus lentus* (SBL, EC 3.4.21.14) is well suited as an exploratory vehicle for evaluating the potential of this combined site-directed mutagenesis chemical modification approach since it is a well-characterized enzyme and is of synthetic (56, 57) as well as industrial (58) interest. Furthermore, SBL's high-resolution crystal structure has been solved (59, 60), and it has been cloned, overexpressed, and purified (61), and its kinetic behavior well characterized (62–65). In addition, and importantly, wild-type (WT) SBL contains no natural cysteine residues, and methanethiosulfonate reagents therefore react *only* with the introduced cysteine residue. The validity of the CMM approach for altering the stability (66), specificity (24), kinetic properties (35–37, 39, 40), and pH profiles (38) of subtilisins has been recognized.

WT-SBL has a marked preference for substrates with large hydrophobic uncharged  $\text{P}_1$  residues. In this study, we explore tailoring of the  $\text{S}_1$  pocket of SBL to also accept small hydrophobic, positively charged and negatively charged  $\text{P}_1$  residues. To achieve this broadened  $\text{P}_1$  tolerance, a simplistic strategy of steric and electrostatic complementarity was applied (67). Employing the crystal structure of SBL as our guide (59), the Ser166 residue, which is located at the bottom of the  $\text{S}_1$  pocket and whose side chain points inward toward the pocket, was chosen for mutagenesis to cysteine and subsequent chemical modification. First, to expand SBL's specificity toward small uncharged  $\text{P}_1$  residues, such as the small  $\text{P}_1$  Ala residue of the suc-AAPA-pNA substrate, we introduced large moieties at position 166 in  $\text{S}_1$ , such as benzyl (-c), decyl (-e), cyclohexyl (-f), and steroidyl (-g) groups with a view to reducing the volume of  $\text{S}_1$  and inducing a better fit of small  $\text{P}_1$  groups, thereby conferring elastase-like (68) substrate specificity on SBL. Then, to expand SBL's specificity toward positively charged  $\text{P}_1$  residues, such as the  $\text{P}_1$  Arg residue of the suc-AAPR-pNA substrate, we introduced negatively charged groups at position S166C in  $\text{S}_1$ , such as the ethylsulfonate (-b) moiety, and the dicarboxylic aromatic (-d) and aliphatic mono- (h), di- (i), and tri (-j)-aliphatic groups, to elicit complementary electrostatic attractions with a view to making SBL trypsin-like in its specificity (69). Conversely, to expand SBL's specificity toward negatively charged  $\text{P}_1$  residues, such as the negatively charged  $\text{P}_1$  Glu residue of the suc-AAPE-pNA substrate, we introduced the positively charged ethylamino (-a) group at position S166C in  $\text{S}_1$ .

## RESULTS

The preparations of the requisite MTS reagents **1c** (36), **1e,f** (36), **1d** (36, 70), and **1i,j** (36, 70) are as reported previously, and the steroidyl MTS reagent **1g** was prepared from cholic acid by the same methodology.

Each of the CMMs obtained was characterized in order to establish its purity and integrity. Titration of the CMMs with Ellman's reagent showed a residual thiol content of less than 2% in all cases, demonstrating that the MTS reactions were virtually quantitative. Mass analyses of the CMMs by electrospray mass spectrometry were consistent ( $\pm 6$  Da) with the calculated masses. The purities of the modified enzymes were assessed by native-PAGE, and in all cases, only one band was visible. Furthermore, as expected relative to WT, the negatively charged CMMs S166C-S-b, -d, and -i to -j displayed retarded mobility in the direction of the cathode, while the positively charged S166C-S-a CMM displayed greater mobility. That modification of cysteine is wholly responsible for altered activity was established by the absence of reaction of WT-SBL with the MTS reagents. Also, the modifications are fully reversible by treatment of each of the CMMs with  $\beta$ -mercaptoethanol, further verifying that chemical modification at cysteine was solely responsible for the observed changes in activity. The total amount of active enzyme was determined by titration with phenylmethanesulfonyl fluoride (71).

Initially, three CMMs S166C-S-a, -b, and -c, with a positive, a negative, and with a large hydrophobic side chain, respectively, were subjected to a  $k_{\text{cat}}/K_M$  screen with each of the test substrates, suc-AAP-F/A/R/E-pNA, in order to

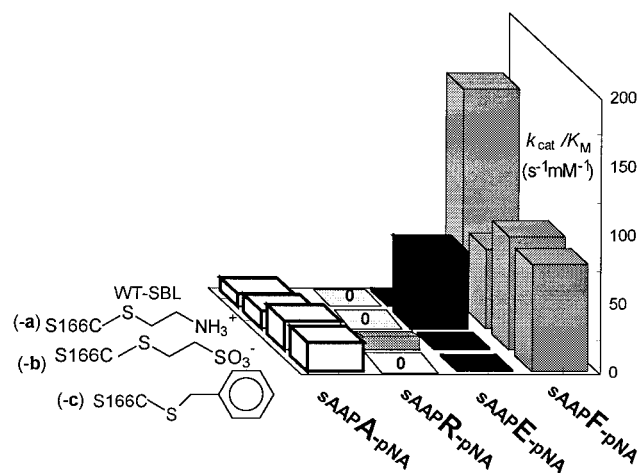


FIGURE 1:  $k_{cat}/K_M$  screen of WT-SBL and S166C-S-a to -c CMMs with each of the standard suc-AAPF-pNA substrate and with the suc-AAP-A/R/E-pNA substrates.

identify any induced complementary electrostatic or improved hydrophobic interactions (Figure 1). While, as expected,  $k_{cat}/K_M$ s with the standard suc-AAPF-pNA were lowered, the  $k_{cat}/K_M$ s of the CMMs whose S<sub>1</sub> sites were tailored toward the Ala, Arg, and Glu P<sub>1</sub> residues, improved with the appropriate substrate. This is illustrated in Figure 1 in the higher activity of S166C-S-c with suc-AAPA-pNA, of S166C-S-b with suc-AAPR-pNA, and of S166C-S-a with suc-AAPE-pNA, all relative to WT.

Following the validation of the general design strategy from this initial screen, more complete kinetic analyses were undertaken. The substrate specificity of each of the CMMs was evaluated kinetically with the standard large hydrophobic P<sub>1</sub> residue containing substrate suc-AAPF-pNA. In addition, the S166C CMMs modified with the large hydrophobic MTS reagents **1c,e-g**, were evaluated with the small hydrophobic P<sub>1</sub> residue containing substrate, suc-AAPA-pNA. The S166C CMMs modified with the negatively charged MTS reagents **1b,d,h-j** were evaluated with the positively charged P<sub>1</sub> residue containing substrate, suc-AAPR-pNA. The S166C CMM modified with the positively charged MTS reagent **1a** was evaluated with the negatively charged P<sub>1</sub> residue containing substrate, suc-AAPE-pNA. The results are summarized in Table 1.

## DISCUSSION

The significant substrate preference of WT-SBL for large hydrophobic P<sub>1</sub> residues is apparent from its preference for the Phe P<sub>1</sub> residue of the standard suc-AAPF-pNA substrate, by a factor of 950-fold over the small P<sub>1</sub> residue of suc-AAPA-pNA, by a factor of 24-fold compared to the positively charged P<sub>1</sub> residue of suc-AAPR-pNA and by a factor of 522-fold compared to the negatively charged P<sub>1</sub> residue of suc-AAPE-pNA (Table 1, entries 1, 12, 17, and 23). These kinetic differences are due to changes in both binding, as reflected by  $K_M$ , and in turnover number,  $k_{cat}$ . Moreover, and predictably, the WT enzyme is by far the best catalyst with suc-AAPF-pNA, and its conversions to any of the CMMs were deleterious with respect to this substrate and resulted in  $k_{cat}/K_M$  decreases of up to 34-fold (Table 1, entries 2–11).

To improve the substrate specificity of SBL toward small hydrophobic P<sub>1</sub> residues such as Ala, the simplistic approach

of filling up the S<sub>1</sub> binding cleft was addressed by preparing the S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub> (-c), S166C-S-CH<sub>2</sub>(CH<sub>2</sub>)<sub>8</sub>CH<sub>3</sub> (-e), S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>11</sub> (-f), and S166C-S-steroidyl (-g) CMMs. This design strategy attempted to mimic the function of the bulky S<sub>1</sub>-pocket side chains of  $\alpha$ -lytic protease (72–74) and of elastase (68), which are responsible for their substantial preference for the small P<sub>1</sub> group containing suc-AAPA-pNA substrate over the large P<sub>1</sub>-group containing suc-AAPF-pNA substrate (68, 72–74). These CMMs (S166C-S-c, -e, -f, -g, Table 1, entries 12–16) were then evaluated with the suc-AAPA-pNA substrate. All revealed slightly improved binding compared to WT, with the greatest improvement in  $K_M$  being 2-fold for the S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub> (-c) CMM. However, of these four CMMs, only S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>11</sub> (-f) showed both an improved  $k_{cat}$  and an improved  $k_{cat}/K_M$ . While this design strategy yielded only one CMM with an increased preference for the small Ala P<sub>1</sub> residue, all of these modifications effectively excluded the larger Phe P<sub>1</sub> residue preferred by WT-SBL (Table 1, entries 4 and 6–8). Overall, the selectivities with respect to  $k_{cat}/K_M$  for the suc-AAPA-pNA substrate compared to the suc-AAPF-pNA substrate were improved by 11-fold for S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub> (-c), 1.1-fold for S166C-S-CH<sub>2</sub>(CH<sub>2</sub>)<sub>8</sub>CH<sub>3</sub> (-e), 51-fold for S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>11</sub> (-f), and 3.2-fold for S166C-S-steroidyl (-g), all compared to WT. These differences in P<sub>1</sub> Ala selectivity may be a reflection of the orientation of the R side chain of the CMM, with the side chains of S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub> (-c) and -CH<sub>2</sub>C<sub>6</sub>H<sub>11</sub> (-f), behaving as though directed into the pocket and favoring P<sub>1</sub> = Ala, whereas the side chains of S166C-S-CH<sub>2</sub>(CH<sub>2</sub>)<sub>8</sub>CH<sub>3</sub> (-e) and S166C-S-steroidyl (-g) behave as though directed outward, thus not significantly altering the shape of the S<sub>1</sub> pocket.

The above improvements in P<sub>1</sub> Ala acceptance, although modest, are encouraging and demonstrate the effectiveness of the adopted strategy (20–22, 75). Tailoring the steric complementarity between enzymes and substrates has already been found to be challenging, and comparison of these CMM results with previous literature studies targeting the same goal are both interesting and intriguing. For example, the G166I mutation of subtilisin BPN' effected an almost 1000-fold decrease in  $k_{cat}/K_M$  with the P<sub>1</sub>=Phe suc-AAPF-pNA substrate. However, this same G166I mutant, which was by far the most effective mutation, elicited a 10-fold improved  $k_{cat}/K_M$  compared to WT with the P<sub>1</sub>=Ala suc-AAPA-pNA substrate (20). Notably, the G166I subtilisin mutant is more selective for Ala over Phe than our most selective CMM. However, in both cases, the increases in selectivity are due mainly to decrease in  $k_{cat}/K_M$  for suc-AAPF-pNA rather than increases in  $k_{cat}/K_M$  for suc-AAPA-pNA. The results for both CMM and SDM approaches agree that decreasing the selectivity of an enzyme for a large hydrophobic residue containing substrate can be accomplished in a relatively facile manner by the introduction of large amino acid in the enzyme pocket, but that increasing the selectivity of an enzyme for a small hydrophobic residue containing substrate is much more difficult. Similarly, the G127A mutant of subtilisin YaB, whose specificity was already elastase-like, effected a 10-fold improvement in  $k_{cat}/K_M$  with the suc-AAPA-pNA substrate (75). However, the G127V mutant of subtilisin E induced a decrease in  $k_{cat}/K_M$  with the suc-AAPA-pNA substrate, identifying an inconsistency in the SDM strategy (21). Thus, the CMM approach offers a complementary



Table 1: Kinetic Evaluation<sup>a</sup> of Altered S<sub>1</sub> Pocket Specificity

entry	enzyme	substrate	$K_M$ (mM)	$k_{cat}$ (s <sup>-1</sup> )	$k_{cat}/K_M$ (s <sup>-1</sup> mM <sup>-1</sup> )
1	WT	suc-AAPF-pNA	0.73 ± 0.08	153 ± 4	209 ± 15
2	S166C-S-a	suc-AAPF-pNA	0.68 ± 0.04	50 ± 1	74 ± 5
3	S166C-S-b	suc-AAPF-pNA	1.34 ± 0.08	25.0 ± 0.7	19 ± 1
4	S166C-S-c	suc-AAPF-pNA	1.17 ± 0.06	23.1 ± 0.5	20 ± 1
5	S166C-S-d	suc-AAPF-pNA	1.6 ± 0.2	47 ± 3	29 ± 4
6	S166C-S-e	suc-AAPF-pNA	1.09 ± 0.07	82 ± 2	75 ± 5
7	S166C-S-f	suc-AAPF-pNA	0.70 ± 0.05	4.8 ± 0.1	6.90 ± 0.05
8	S166C-S-g	suc-AAPF-pNA	0.74 ± 0.07	29 ± 1	41 ± 4
9	S166C-S-h <sup>b</sup>	suc-AAPF-pNA	1.52 ± 0.06	48 ± 1	31 ± 1
10	S166C-S-i <sup>b</sup>	suc-AAPF-pNA	2.26 ± 0.10	67 ± 2	30 ± 2
11	S166C-S-j <sup>b</sup>	suc-AAPF-pNA	2.46 ± 0.11	76 ± 2	31 ± 2
12	WT	suc-AAPA-pNA	2.0 ± 0.1	17.7 ± 0.3	8.8 ± 0.4
13	S166C-S-c	suc-AAPA-pNA	0.8 ± 0.1	6.8 ± 0.3	9 ± 1
14	S166C-S-e	suc-AAPA-pNA	1.90 ± 0.03	6.8 ± 0.4	3.6 ± 0.6
15	S166C-S-f	suc-AAPA-pNA	1.90 ± 0.07	28.2 ± 0.4	14.8 ± 0.6
16	S166C-S-g	suc-AAPA-pNA	1.74 ± 0.04	9.65 ± 0.07	5.54 ± 0.3
17	WT	suc-AAPR-pNA	7.2 ± 0.7	0.16 ± 0.01	0.022 ± 0.002
18	S166C-S-b	suc-AAPR-pNA	3.4 ± 0.3	0.17 ± 0.01	0.050 ± 0.005
19	S166C-S-d	suc-AAPR-pNA	5.5 ± 1.1	0.68 ± 0.08	0.12 ± 0.03
20	S166C-S-h	suc-AAPR-pNA	8.2 ± 0.9	0.35 ± 0.02	0.041 ± 0.005
21	S166C-S-i	suc-AAPR-pNA	5.3 ± 0.5	0.43 ± 0.02	0.080 ± 0.008
22	S166C-S-j	suc-AAPR-pNA	5.2 ± 0.6	1.06 ± 0.07	0.20 ± 0.03
23	WT	suc-AAPE-pNA	4.4 ± 0.4	1.75 ± 0.08	0.40 ± 0.04
24	S166C-S-a	suc-AAPE-pNA	1.9 ± 0.1	14.5 ± 0.3	7.6 ± 0.4

<sup>a</sup> Michaelis–Menten constants were measured by the initial rates method in, pH 8.6, Tris-HCl buffer at 25 °C with suc-AAPF-pNA as the substrate. <sup>b</sup> Taken from ref 70.

alternative to conventional site-directed mutagenesis toward the goal of tailoring the steric complementarity between enzymes and substrates (22).

Improving the substrate specificity of SBL toward positively charged P<sub>1</sub> residues such as Arg was based on mimicking the common motif in trypsin-like enzymes (69) of high negative charge density of acidic residues that favor binding of positively charged substrate structures (69, 76–78). This goal was addressed by S166C-S-CH<sub>2</sub>CH<sub>2</sub>SO<sub>3</sub><sup>-</sup> (-b), S166C-S-CH<sub>2</sub>(CH<sub>2</sub>)<sub>2</sub>CH<sub>2</sub>COO<sup>-</sup> (-h), S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>4</sub>-3,5-(COO<sup>-</sup>)<sub>2</sub> (-d), S166C-S-CH<sub>2</sub>CH<sub>2</sub>C(CH<sub>3</sub>)(COO<sup>-</sup>)<sub>2</sub> (-i), and S166C-S-CH<sub>2</sub>CH<sub>2</sub>C(COO<sup>-</sup>)<sub>3</sub> (-j), a series of CMMs which provide an S<sub>1</sub> pocket which is potentially mono-, di-, and trinegatively charged. Evaluation of each of these CMMs with the suc-AAPR-pNA substrate revealed  $K_M$ s that were up to 2-fold improved compared to WT (Table 1, entries 17–22). The general success of this approach is evident since all of the CMMs with a negatively charged -R side chain showed improved activity compared to WT with an up to 7-fold improved  $k_{cat}$  and an up to 9-fold improved  $k_{cat}/K_M$  with the suc-AAPR-pNA substrate (Table 1, entries 17–22). Overall, the selectivities, with respect to  $k_{cat}/K_M$ , for the suc-AAPR-pNA substrate compared to the suc-AAPF-pNA substrate were improved 25-fold for S166C-S-CH<sub>2</sub>CH<sub>2</sub>SO<sub>3</sub><sup>-</sup> (-b), 13-fold for S166C-S-CH<sub>2</sub>(CH<sub>2</sub>)<sub>2</sub>CH<sub>2</sub>COO<sup>-</sup> (-h), 39-fold for S166C-S-CH<sub>2</sub>C<sub>6</sub>H<sub>4</sub>-3,5-(COO<sup>-</sup>)<sub>2</sub> (-d), 25-fold for S166C-S-CH<sub>2</sub>CH<sub>2</sub>C(CH<sub>3</sub>)(COO<sup>-</sup>)<sub>2</sub> (-i), and 61-fold for S166C-S-CH<sub>2</sub>CH<sub>2</sub>C(COO<sup>-</sup>)<sub>3</sub> (-j) relative to WT.

The strategy of introducing charge complementarity to induce trypsin-like P<sub>1</sub> specificity in subtilisins has previously been explored using site-directed mutagenesis (SDM) (12, 16, 79). Interestingly, however, the G166D and G166E mutants of subtilisin BPN' caused decreases in  $k_{cat}/K_M$  with the suc-AAPK-pNA substrate rather than the anticipated increases (12). Furthermore, while the G166D subtilisin BPN' mutant was reported to exhibit a 18-fold improvement in

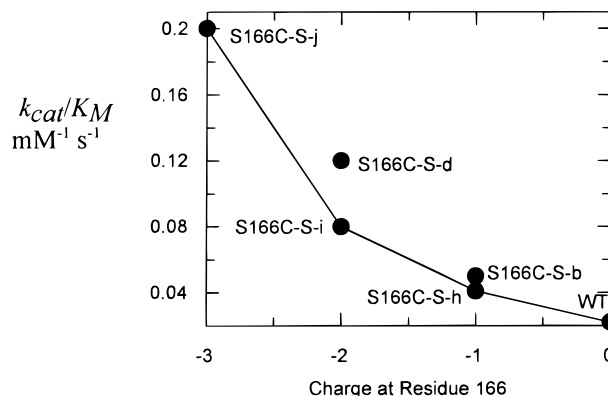


FIGURE 2: Plot of side-chain negative charge on R at position 166 versus  $k_{cat}/K_M$  with suc-AAPR-pNA. The points on the line are for the aliphatic series of carboxylated -R groups and the WT, Ser166.

Arg/Phe P<sub>1</sub> selectivity, this was accompanied by a 2.5-fold decrease in  $k_{cat}/K_M$  with the suc-AAPR-pNA substrate compared to WT (16). Thus, compared to WT, both with respect to improved  $k_{cat}/K_M$  with the suc-AAPR-pNA substrate and improved Arg/Phe P<sub>1</sub> selectivity, the CMMs reported herein are more successful.

Furthermore, these improvements correlate directly with the number of introduced negative charges, such that each additional negative charge introduced at position 166 causes an approximate doubling in  $k_{cat}/K_M$  with the complementary charged suc-AAPR-pNA substrate (Figure 2). This arithmetic improvement is interesting since it contrasts with the previous observation that while remote noninteracting charged mutations are additive, multiple interacting adjacent charged mutations often are not additive. Rather, in most cases, the empirically observed effects are lower than predicted from their individual sums (80). This phenomenon is particularly problematic for charged mutations due to the long-range effects of electrostatic interactions. For example, the sum of the transition-state stabilization energy,  $\Delta\Delta G^\ddagger$ , for the two

single positively charged subtilisin mutations (D99K and E156K) over estimates the empirically observed effect of the double mutant, when assayed with an Arg P<sub>1</sub> containing substrate (80). In contrast, in all cases the empirically determined  $\Delta\Delta G^\ddagger_T$  values for the aliphatic carboxylate series of mono- di- and trinegatively charged side chains, of the S166C-S-**h**, -**i**, and -**j** CMMs exhibit an additive effect of additional charge (81). Thus, the CMM approach offers a convenient method to circumvent the problem of the attenuation of the augmenting effect of the introduction of additional charges by SDM by permitting the introduction of a larger local charge density.

Conversely, the adopted design strategy of introducing a complementary positive charge in the S<sub>1</sub> binding cleft by the CMM approach to improve P<sub>1</sub> = Glu selectivity was based on mimicking the specificity determinants of the serine proteases Pronase (82–83) and granzyme B (84–86), which exhibit a substrate preference for negatively charged P<sub>1</sub> residues, and whose S<sub>1</sub> pockets are lined with positively charged residues. The success of the current approach is apparent from the remarkable 19-fold increase in  $k_{cat}/K_M$ , with the suc-AAPE-pNA substrate displayed by S166C-S-CH<sub>2</sub>-CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> (-**a**). This enhancement is due to a combination of better binding, evident from the 2-fold lower  $K_M$  and 8-fold higher  $k_{cat}$  (Table 1, entries 23 and 24). The induction of electrostatic complementarity was most unequivocally demonstrated by the 54-fold improvement in suc-AAPE-pNA to suc-AAPF-pNA substrate selectivity, with respect to  $k_{cat}/K_M$ , for S166C-S-CH<sub>2</sub>CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> (-**a**) compared to WT. Previously, the individual G166R and G166K subtilisin BPN' mutations elicited 23- and 340-fold improvements in  $k_{cat}/K_M$  for the suc-AAPE-pNA substrate (12, 14). However, it must be noted that the E156Q–G166K double mutant was much more receptive to Glu P<sub>1</sub> and exhibited a 1900-fold improvement compared to WT (12). Interestingly, both the G166R and G166K mutants displayed even higher  $k_{cat}/K_M$ s with the hydrophobic P<sub>1</sub> residue containing substrates suc-AAPN-pNA and suc-AAPM-pNA and even with the positively charged P<sub>1</sub> residue containing substrate suc-AAPK-pNA (12). While S166C-S-CH<sub>2</sub>CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> (-**a**) still exhibits a 10-fold preference for suc-AAPF-pNA compared to suc-AAPE-pNA (Table 1, entry 2), the substrate screen (Figure 1) shows that suc-AAPA-pNA and suc-AAPR-pNA are poorer substrates.

Since the S166C-S-CH<sub>2</sub>CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> (-**a**) and suc-AAPE-pNA CMM-substrate pair exhibited the greatest  $k_{cat}/K_M$  improvement relative to WT, at 19-fold (Table 1, entry 24), more detailed insights into the molecular basis of their interaction was sought using molecular modeling. Using the modeling approach reported previously (37), the product inhibitor, AAPE bound to WT-SBL and to the S166C-S-CH<sub>2</sub>CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> (-**a**) CMM was minimized. Molecular modeling revealed that the minimized binding conformations of AAPE to both the WT and S166C-S-CH<sub>2</sub>CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> enzymes are quite similar despite the 19-fold difference in  $k_{cat}/K_M$ . However, the ammonium moiety of the S166C-S-CH<sub>2</sub>CH<sub>2</sub>-NH<sub>3</sub><sup>+</sup> CMM side chain is oriented toward the carboxylate of the glutamic acid P<sub>1</sub> residue, and although it is not quite within salt-bridge distance (N<sup>+</sup> -to- <sup>-</sup>OOC, 4.76 Å), this additional favorable Coulombic interaction between the ammonium side chain of S166C-S-CH<sub>2</sub>CH<sub>2</sub>NH<sub>3</sub><sup>+</sup> and the carboxylate of the glutamic acid P<sub>1</sub> residue is deemed

responsible for the observed 19-fold improvement in  $k_{cat}/K_M$  for this CMM-substrate pair, compared to WT.

## CONCLUSION

While much remains to be done, the overall data clearly provides an encouraging initial validation of the practical effectiveness of the CMM strategy for generating complementary electrostatic and steric enzyme–substrate interactions. For each of the Ala, Arg, and Glu P<sub>1</sub> residues, at least one, and up to five, of the designed CMMs exhibit improved  $k_{cat}/K_M$ s compared to WT. The CMM approach is complementary to the SDM approach and also offers the additional opportunity for the introduction of multiply charged side chains generating high charge densities at single active-site locations. The beneficial effects of the introduction of a localized high charge density was demonstrated by the arithmetic increases in  $k_{cat}/K_M$ , with the suc-AAPR-pNA substrate, induced by incremental increases in the negative charge of the S<sub>1</sub> pocket. Further studies toward refining this general strategy are in hand and will be reported in due course.

## EXPERIMENTAL SECTION

Sulfonatoethyl methanethiosulfonate (**1a**) and ethylammonium methanethiosulfonate (**1b**) were purchased from Toronto Research Chemicals (2 Brisbane Rd., Toronto, ON, Canada). Reagents **1c–f** (36) and **1h–j** (70) were prepared as previously described. ES-MS data were acquired using a PE SCIEX API III Biomolecular mass spectrometer. The tetrapeptide substrates suc-AAPF/A/R/E-pNA were purchased from Bachem Bioscience Inc. (Torrance, CA). All buffer solutions were made up in deionized water.

**Site-Specific Chemical Modification.** To 25 mg of a S166C mutant, purified as previously described (37, 61) and stored flash frozen in CHES buffer (2.5 mL; 70 mM Ches, 5 mM Mes, and 2 mM CaCl<sub>2</sub>, pH 9.5) at 20 °C, was added one of the methanethiosulfonate reagents (**1a–g**) (100 µL of a 0.2 M solution), in a PEG (10 000)-coated polypropylene test tube, and the mixture agitated in an end-over-end rotator. Blank reactions containing 100 µL of solvent instead of the reagent solution were run in parallel. Each of the modification reactions was monitored spectrophotometrically ( $\epsilon_{410} = 8800 \text{ M}^{-1} \text{ cm}^{-1}$ ) (79) on a Perkin-Elmer Lambda 2 spectrophotometer, by specific activity measurements. After the reaction was quenched by dilution in MES buffer (5 mM Mes and 2 mM CaCl<sub>2</sub>, pH 6.5) at 0 °C, the specific activity of the CMM (10 µL), was determined in buffer containing 0.1 M Tris, pH 8.6, 0.005% Tween 80, and 1% DMSO, with the suc-AAPF-pNA substrate (1 mg/mL) at 25 °C. The reaction was terminated when the addition of a further 100 µL of methanethiosulfonate solution effected no further change in specific activity, generally in 30 min to 3 h. The reaction solution was purified on a disposable desalting column (Pharmacia Biotech PD-10, Sephadex G-25 M) preequilibrated with Mes buffer (5 mM Mes and 2 mM CaCl<sub>2</sub>, pH 6.5) then dialyzed against 20 mM Mes and 1 mM CaCl<sub>2</sub>, pH 5.8 (3 × 1 L) at 4 °C and aliquoted into 0.5–1.5 mL volumes, flash frozen in liquid nitrogen, and then stored at –20 °C. Modified enzymes were analyzed by nondenaturing gradient (8 to 25%) gels at pH 4.2, run toward the cathode on the Pharmacia Phast-System, and appeared as one single band.

**Electrospray Mass Spectrometry.** Prior to ES-MS analysis, CMMs were purified by FPLC (Bio-Rad, Biologic System) on a Source 15 RPC matrix (17-0727-20 from Pharmacia) with 5% acetonitrile, 0.01% TFA as the running buffer and eluted with 80% acetonitrile, 0.01% TFA in a one step gradient. (Mass) WT: calcd, 26 698; found, 26 694 (36). S166C-S-a: calcd, 26 714; found, 26 708 (36). S166C-S-b: calcd, 26 853; found, 26 851 (36). S166C-S-c: calcd, 26 836; found, 26 832 (36). S166C-S-d: calcd, 26 924; found, 26 928 (70). S166C-S-e: calcd, 26 886; found, 26 890 (36). S166C-S-f: calcd, 26 842; found, 26 844 (36). S166C-S-g: calcd, 27 128; found, 27 123 (36). S166C-S-h: calcd, 26 846; found, 28 646 (70). S166C-S-i: calcd, 26 890; found, 26 894 (70). S166C-S-j: calcd, 26 934; found, 26 939 (70).

**Regeneration of Unmodified Enzyme by Treatment with  $\beta$ -Mercaptoethanol.** To a solution of CMM (2.0 mg) in 250  $\mu$ L of Ches buffer (70 mM Ches, 5 mM Mes, and 2 mM  $\text{CaCl}_2$ , pH 9.5) was added 10  $\mu$ L of a solution of  $\beta$ -mercaptoethanol (1 M in 95% EtOH). The reaction was monitored by specific activity measurements.

**Free Thiol Titration.** The free thiol content of S166C CMMs was determined spectrophotometrically by titration with Ellman's reagent ( $\epsilon_{412} = 13\,600\text{ M}^{-1}\text{ cm}^{-1}$ ) (87) in phosphate buffer 0.25 M, pH 8.0.

**Active-Site Titrations.** The active enzyme concentration was determined as previously described (71) by monitoring fluoride release upon enzyme reaction with phenylmethanesulfonyl fluoride (Aldrich Chemical Co. Inc.) as measured by a fluoride ion sensitive electrode (Orion Research 96-09). The active enzyme concentration determined in this way was used to calculate kinetic parameters for each CMM.

**Kinetic Measurements.** Michaelis–Menten constants were measured at 25 °C by curve fitting (GraFit 3.03) of the initial rate data determined at eight concentrations (0.125–8.0 mM) of the suc-AAPX-pNA substrate in, pH 8.6, 0.1 M Tris-HCl buffer containing 0.005% Tween 80 and 1% DMSO ( $\epsilon_{410} = 8800\text{ M}^{-1}\text{ cm}^{-1}$ ).

**Molecular Modeling.** The X-ray structure of subtilisin *Bacillus lentus* (59) was used as the starting point for calculations on the wild-type and chemically modified mutant enzymes. The enzyme setup was performed with Insight II (88). To create initial coordinates for the minimization, hydrogens were added at the pH used for kinetic measurements. This protonated all Lys and Arg residues and the N-terminus and deprotonated all Glu and Asp residues and the C-terminus. In addition, the active-site His64 was protonated. The model system with the Ala-Ala-Pro-Phe (from crystal structure) (59) product inhibitor bound in the  $S_1$ – $S_4$  pocket was solvated with a 5 Å layer of water molecules giving a total number of water molecules of 1143 in this system. The overall charge of the enzyme–inhibitor complex resulting from this setup was +4 for the WT enzyme. Energy simulations were performed with the Discover program (89), on a Silicon Graphics Iris Indigo computer, using the consistent valence force field function (CVFF). A nonbonded cutoff distance of 18 Å with a switching distance of 2 Å was employed. The nonbonded pair list was updated every 20 cycles, and a dielectric constant of 1 was used in all calculations. The WT enzyme was minimized in stages, with initially only the water molecules being allowed to move, then the water molecules and the amino acid side chains, and then the entire enzyme. The mutated and chemically modified enzymes were gener-

ated using the Builder module of Insight. Then the amino acid side chains within a 10 Å radius of the  $\alpha$ -carbon of the mutated residue were minimized while all other residues were constrained, then all of the atoms within a 10 Å shell were minimized, followed by minimization of the whole system. To examine the effect of a different  $P_1$  residue (Glu), the Phe to Glu mutation of the product inhibitor was constructed using insightII, and then this structure was minimized as above.

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## SUPPORTING INFORMATION AVAILABLE

Preparation of steroidyl methane thiosulfonate reagent, **1g**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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81.  $\Delta\Delta G^\ddagger T = RT \ln[(k_{\text{cat}}/K_M)_{\text{WT}}/(k_{\text{cat}}/K_M)_{\text{CMM}}]$ ; S166C-S-b,  $-2.04$  kJ mol $^{-1}$ ; S166C-S-d,  $-4.21$  kJ mol $^{-1}$ ; S166C-S-h,  $-1.54$  kJ mol $^{-1}$ ; S166C-S-i,  $-3.20$  kJ mol $^{-1}$ ; and S166C-S-j,  $-5.47$  kJ mol $^{-1}$ .
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